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水体中短链氯化石蜡污染研究进展*

Review on Short-chain Chlorinated Paraffins Pollution in Water

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摘要:短链氯化石蜡(SCCPs)拟列入“斯德哥尔摩公约”禁止使用的一类持久性有机污染物,对生态环境和人体健康具有极大的潜在危害,已成为国际环境科学领域的一个研究热点。水体是SCCPs重要的“汇”,本文着重综述了水生态系统中沉积物、水、生物体等不同环境介质中SCCPs的含量分布及对水生生物的危害,并对SCCPs的研究重点和前景进行展望。

关键词:短链氯化石蜡 水体 污染状况

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Abstract: Short-chain chlorinated paraffins (SCCPs), as a candidate of persistent organic pollutants in “Stockholm Convention”, have potential hazard to ecological environment and people health, which becomes a new hotspot in field of environmental science worldwide. The water is an important destination of SCCPs. The distribution of SCCPs in different environmental media and the potential threaten of SCCPs to aquatic animals were critically reviewed in this paper. Meanwhile, the research progress and trend of SCCPs were also reviewed.

Key words: short-chain chlorinated paraffins, water, contamination level

0 引言

氯化石蜡(CPs)是直链正构烷烃的氯代衍生物,工业上用作阻燃剂、增塑剂、金属加工油和皮革处理剂等,其中碳链长度为10~13个碳原子的氯化石蜡被定义为短链氯化石蜡(SCCPs)。相对于中链氯化石蜡(MCCPs,碳链长度为14~17)和长链氯化石蜡(LCCPs,碳链长度为18~30),短链氯化石蜡因其较强的生物累积特性和更高的毒性而受到广泛

关注。自20世纪50年代以来,我国已经开始生产氯化石蜡,随着国内塑料制品工业的迅速发展,氯化石蜡行业发展迅速。目前,我国已是氯化石蜡第一生产大国,截至2015年,国内氯化石蜡总产能达160万t。氯化石蜡产品中SCCPs含量普遍在6%以上^[1-2]。这一方面对我国的生态环境和人体健康带来极大的潜在危害,另一方面面临着国际社会要求淘汰SCCPs消费的巨大压力。

2002年,欧盟水框架指令已禁止使用SCCPs,并将其列为水中的首要危险化学品之一。2008年“关于持久性有机污染物(POPs)斯德哥尔摩公约”将其列入POPs备选清单,我国也于同年启动对SCCPs的评估工作。2011年的斯德哥尔摩大会上SCCPs仅以一票之差最终未被列入POPs受控名

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单^[3]。2015年3月欧盟对现行的塑料制品法规(EU)No10/2011进行升级,法规中进一步要求减少SCCPs的使用量^[4]。2015年11月欧盟又在《持久性有机污染物法规》中修订了对SCCPs的限制^[5]。现有研究表明,SCCPs普遍存在于各种环境介质中,尤其在水体中广泛存在。本文综述了SCCPs在水体中的环境行为及毒性研究进展,期望对SCCPs的深入研究提供参考。

1 SCCPs的理化性质

SCCPs是一类组成复杂的混合物,常温下为淡黄色或无色粘稠液体。SCCPs亨利定律常数是 $0.7\sim 18\text{ Pa}\cdot\text{m}^3\cdot\text{mol}^{-1}$,蒸汽压为 $3.2\times 10^{-4}\sim 0.066\text{ Pa}$ (文献^[6]),这表明SCCPs可以从大气中沉降到土壤、底泥和水中,也可以从上述介质中再活化到空气中。SCCPs的辛醇水分配系数(LogK_{ow})的值为 $4.39\sim 8.69$ (文献^[7]),具有较高的脂溶性,因此在水生生物中有较强的生物富集及放大能力。SCCPs也具有较高的辛醇空气分配系数(logK_{OA}),表明SCCPs具有吸附在大气的有机相中进行长距离迁移的能力^[8]。多数SCCPs混合物的溶解度是 $22.4\sim 994\ \mu\text{g}\cdot\text{L}^{-1}$, C_{10} 和 C_{13} 混合物的溶解度为 $6.4\sim 2370\ \mu\text{g}\cdot\text{L}^{-1}$ (文献^[6])。随着碳链的增加,SCCPs的水溶性和蒸气压均有下降的趋势,但与一些氯化芳烃不同,在5个氯原子以内,随着氯原子个数的增加,水溶性增大^[9]。

2 水体中SCCPs的污染现状

目前为止还未发现自然来源的SCCPs。水体中SCCPs主要来自SCCPs及其他链长CPs生产、储存、运输和使用过程中的排放;设备冲洗和使用后的金属加工/切削液也都可能成为SCCPs进入水生生态系统的源头。

2.1 SCCPs在沉积物中的分布

沉积物是SCCPs重要的“汇”,是国内外SCCPs研究的热点领域。在欧洲工业发达地区,巴塞罗那河流底泥中的含量为 $0.25\sim 3.04\ \mu\text{g}\cdot\text{g}^{-1}$ (干重)^[10];挪威河流沉积物中检测出很高的SCCPs($19400\text{ ng}\cdot\text{g}^{-1}$)(湿重)^[11]。捷克工业区附近河流沉积物中SCCPs浓度介于无检测值至 $347.4\text{ ng}\cdot\text{g}^{-1}$ (干重)之间^[12]。北美五大湖地区,在安大略湖汉米尔顿港附近的沉积物中检测到最高浓度SCCPs,五大湖其他区域SCCPs浓度为 $5.9\sim 410\text{ ng}\cdot\text{g}^{-1}$ (干重)^[13-14]。亚洲地区,日本2条河流沉

积物中SCCPs水平为 $4.9\sim 484.4\text{ ng}\cdot\text{g}^{-1}$ (干重)^[15]。上述研究表明人类工业活动的强弱是决定该地区沉积物中SCCPs含量的主要原因。Tomy等^[16-17]检测到北极地区偏远湖泊沉积物中的SCCPs浓度为 $1.6\sim 257\text{ ng}\cdot\text{g}^{-1}$ (干重),发现SCCPs在包括极地区域在内的广泛地理范围内普遍存在,表明SCCPs可以通过长距离大气传输在全球范围内分布。我国的SCCPs研究起步较晚,但进展很快。陈满英等^[18]报道珠江三角洲地区沉积物(池塘、河流和河口及柱芯)SCCPs的浓度为 $320\sim 6600\text{ ng}\cdot\text{g}^{-1}$ (干重),该研究发现SCCPs沉积埋藏过程中可能存在脱氯的降解过程。陈茹等^[19]报道了黄渤海地区沉积物样品中SCCPs浓度在 $14.5\sim 85.2\text{ ng}\cdot\text{g}^{-1}$ (干重),陈晨等^[20]报道辽河口海域沉积物中SCCPs的浓度为 $64.9\sim 1683.4\text{ ng}\cdot\text{g}^{-1}$ (干重)。上述结果暗示河流排放是海洋中SCCPs的主要来源。李慧娟等^[21]报道SCCPs在长江三角洲和闽浙沿岸的表层沉积物含量水平为 $9.0\sim 37.2\text{ ng}\cdot\text{g}^{-1}$ (干重),该研究同时发现SCCPs在沉积物中的迁移很大程度上受总有机碳(TOC)影响。

2.2 SCCPs在水中的分布

虽然SCCPs在水中的溶解度不高,但广泛存在于各种水环境中。西班牙Llobregat河和英国Dawen河的SCCPs浓度分别为 $300\sim 2100\text{ ng}\cdot\text{L}^{-1}$ (文献^[22-23])和 $200\sim 1700\text{ ng}\cdot\text{L}^{-1}$ (文献^[24]),高于已报道的日本几条河流($7.6\sim 220\text{ ng}\cdot\text{L}^{-1}$)^[25-26]及加拿大劳伦斯河($15.74\sim 59.57\text{ ng}\cdot\text{L}^{-1}$)^[27]的SCCPs含量。2000—2004年对安大略湖的持续监测显示,湖中SCCPs的平均浓度为 $1.194\text{ ng}\cdot\text{L}^{-1}$,处于较低水平^[28]。Castells等^[29]对西班牙污水处理厂进出水样品的检测发现:进水口SCCPs浓度为 $310\sim 620\text{ ng}\cdot\text{L}^{-1}$;出水口SCCPs未检出。Zeng等^[30]测得北京高碑店污水处理厂进水口的SCCPs浓度($4200\sim 4700\text{ ng}\cdot\text{L}^{-1}$)显著高于出水口浓度($364\sim 416\text{ ng}\cdot\text{L}^{-1}$),但活性污泥中SCCPs含量为 $(15.6\pm 1.4)\mu\text{g}\cdot\text{g}^{-1}$ (干重),结果暗示虽然污水处理厂能够有效净化水中SCCPs,但沉积物容易造成二次污染。有关海水中SCCPs浓度的报道较少,于国龙^[31]对辽宁普兰店湾的海水进行调查,测得SCCPs的浓度为 $493.87\sim 1490\text{ ng}\cdot\text{L}^{-1}$;马新东等^[32]研究辽东湾海域海水中SCCPs的浓度在 $4.1\sim 13.1\text{ ng}\cdot\text{L}^{-1}$,结果表明海水中SCCPs的污染浓度与人类生产活动正相关。

2.3 SCCPs 在水生生物中的分布

SCCPs 的物理性质决定其在水生生物中有较强的生物富集及放大能力。目前已报道的各种水生生物体内 SCCPs 数据较多。英国河流中,底栖生物、蚯蚓和鱼类体内 SCCPs 的含量为 $0.05 \sim 0.8 \mu\text{g} \cdot \text{g}^{-1}$ (湿重)^[24]。北美五大湖地区,安大略湖鱼类体内的 SCCPs 浓度为 $7.01 \sim 2630 \text{ ng} \cdot \text{g}^{-1}$ (湿重)^[33];密歇根湖和安大略湖糠虾体内 SCCPs 浓度的平均值为 $2.1 \sim 7.5 \text{ ng} \cdot \text{g}^{-1}$ (脂重)^[34];五大湖的鱼类和无脊椎生物中 SCCPs 浓度的平均值为 $130 \sim 500 \text{ ng} \cdot \text{g}^{-1}$ (脂重)^[35]。海洋中,挪威境内鲑鱼、鳕鱼和北极点红鲑等海洋鱼类体内 SCCPs 的浓度在 $108 \sim 3700 \text{ ng} \cdot \text{g}^{-1}$ (脂重)^[11]。Tomy 等^[8]测得北极地区白鲸、环斑海豹和海象等海洋哺乳动物体内的 SCCPs 浓度为 $95 \sim 626 \text{ ng} \cdot \text{g}^{-1}$ (湿重)。上述水生生物样本体内 SCCPs 的主要成份是碳链长度较短和低氯代组分,这些组分的挥发性较高,更易被生物体累积。马新东等^[36]测得辽东湾不同生物体内 SCCPs 的浓度为 $86 \sim 4400 \text{ ng} \cdot \text{g}^{-1}$ (湿重),且不同物种间的含量差异比较显著,其中浓度最低的生物为浮游动物,浓度最高的为扇贝。于国龙等^[31]测得辽河入海口海域 14 种生物体内的 SCCPs 含量为 $758.62 \sim 17027.26 \text{ ng} \cdot \text{g}^{-1}$ (干重)。姜国等^[37]测得长三角和珠三角地区不同种鱼类体内的 SCCPs 含量分别为 $8 \sim 130 \text{ ng} \cdot \text{g}^{-1}$ 和 $9 \sim 118 \text{ ng} \cdot \text{g}^{-1}$ (湿重)。Yuan 等^[38]调查了中国环渤海区域 9 个沿海城市软体动物样品中的 SCCPs 浓度水平,发现 SCCPs 在我国海产品中可普遍被检出。现有研究表明,我国水生生物体内 SCCPs 的质量分数普遍位于目前世界报道水生动物体内 SCCPs 质量分数范围的中高端,赋存程度不容乐观。

2.4 SCCPs 对水生生物毒性效应

目前,国外以大鼠为对象对 SCCPs 的生物毒性做了一些研究。不同剂量水平的 SCCPs 会导致大鼠出现各种临床症状,如脊椎突出,呼吸异常,活动减少,直至死亡^[39]。Cooley 等^[40]发现幼年虹鳟鱼在含有 SCCPs 的环境中暴露 21 d 即出现严重的肝脏组织病理学反应,纤维损害和肝脏细胞坏死严重。Madeley 和 Maddock^[41]通过 SCCPs 对虹鳟鱼的毒性研究,认为 SCCPs 对水生生物的毒性可能需要很长一段时间才能表达出来。另有研究表明,大型蚤 (*Daphnia magna*)、糠虾 (*Mysidopsis bahia*) 和贻贝 (*Mytilus edulis*) 等水生无脊椎动物对 SCCPs 相对比较敏感^[42-43]。现有研究认为

SCCPs 对陆生脊椎动物毒性较低,对鱼类的急性毒性阈值超过其在水中的溶解度,但一些水生无脊椎动物对 SCCPs 则相对比较敏感^[9,44]。另外,由于 SCCPs 是一类亲脂性化合物,储存于脂肪中的 SCCPs 可通过脂肪代谢再进入血液而威胁健康^[45]。总的来说,虽然 SCCPs 对多种生物的毒性作用已有相关报道,但研究广度和深度仍然不足。

3 展望

我国是 CPs 生产和消费大国,也是受 SCCPs 污染较重的国家之一。由于 SCCPs 具有持久性、长距离迁移性、生物富集性和生物毒性等特性,必须关注其对生态环境及人体健康的潜在风险;加大 SCCPs 的研究力度,对控制污染、保护环境具有深远的意义。目前,国内外对 SCCPs 的研究还集中在实验室阶段,不同实验室对同一种 SCCPs 样品分析结果差异巨大,开展相对成熟的、标准化的定量方法研究以及国际实验室间分析比对研究是必要的。SCCPs 毒性效应和致毒机制的研究仍比较粗浅,需加强对其致毒效应的机制和分子机理的研究。另外海洋作为 SCCPs 重要的汇,应加强 SCCPs 在海洋中环境行为的研究力度。

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